NOTES

A Neuronal Cell Protecting Substance, 4-Demethoxymichigazone, Produced by Streptomyces halstedii

Toshihiro Kunigami, Kazuo Shin-ya, Kazuo Furihata[†], Keiko Furihata, Yoichi Hayakawa and Haruo Seto*

Institute of Cellular and Molecular Biosciences, The University of Tokyo, Bunkyo-ku, Tokyo 113, Japan [†]Division of Agriculture and Agricultural Life Sciences, The University of Tokyo, Bunkyo-ku, Tokyo 113, Japan

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During and after brain ischemia, neuronal degeneration in the hippocampus is mediated largely by an excitatory amino acid, L-glutamic acid. Thus, brain ischemia injury may be expected to be overcome by L-glutamate toxicity suppressors. In the course of our screening program for substances which protect neuronal hybridoma N18-RE-105 cells^{1~3)} from L-glutamate toxicity, we isolated carquinostatin A⁴⁾, lavanduquinocin⁵⁾, and aestivophoenins A and B⁶⁾. Further investigation has resulted in the isolation of 4-demethoxymichigazone(8-hydroxymethyl-2-methoxy-3*H*-phenoxazine-3-one) (1) from *Streptomyces halstedii* 2832-SVS6. We report herein the isolation, structure, and biological activities of 1.

Streptomyces halstedii 2832-SVS6, which was isolated from a soil sample collected in Atami, Shizuoka Prefecture, Japan, was cultivated at 27°C in a 60-liter jar fermenter containing 30 liters of a medium consisting of starch 2.5%, soybean meal 1.5%, dry yeast 0.2%, and CaCO₃ 0.4%, pH adjusted to 6.2 before sterilization. After fermentation for 72 hours, the culture filtrate was adsorbed on Diaion HP-20. After washing with 50% MeOH, the active fraction was eluted with 100% MeOH and concentrated. The aqueous residue was extracted with EtOAc, and the organic layer was, after drying over Na₂SO₄, evaporated to give an oily residue. This material

Fig. 1. Structures of 4-demethoxymichigazone (1) and michigazone (2).



4-demethoxymichigazone (1): R = Hmichigazone (2) : $R = OCH_3$ was subjected to silica gel column chromatography developed with a solvent system of *n*-hexane-EtOAc (1:4). The active fraction was then applied to an ODS column (Senshu ODS-SS-1020T) and developed with 60% MeOH. The combined active fraction was purified by preparative HPLC (YMC D-ODS-7) with 20% CH₃CN to give two phenoxazone derivatives, **1** as an orange powder (5.5 mg) and michigazone⁷⁾ (**2**) (1.8 mg) as a minor congener (Fig. 1).

The physico-chemical properties of **1** were as follows: MP 193~194°C; IR v_{max} (KBr) cm⁻¹ 3450, 1620; UV λ_{max} nm (ε) 222 (17,600), 251 (10,600), 390 (11,400), 444 (5,900). The molecular formula of **1** was established by HRFAB-MS as C₁₄H₁₁NO₄ (*m*/*z* 258.0813 (M+H)⁺, calcd 258.0767). The UV and visible absorptions of **1** suggested the presence of a phenoxazone nucleus^{7,8}.

The ¹H and ¹³C NMR assignments of **1** and **2**, which were made by heteronuclear multiple quantum coherence (HMQC) experiments, are summarized in Table 1. The ¹H and ¹³C NMR spectral data revealed the presence of a 1,2,4-trisubstituted benzene moiety. The HMBC spectrum showed a hydroxymethyl proton 11-H (4.69 ppm) long-range coupled to C-7 (130.1 ppm), C-8 (139.1 ppm) and C-9 (127.0 ppm). Thus, the hydroxymethyl residue was located at the C-8 position. Assignments of quaternary carbons in the trisubstituted benzene moiety, C-5a (142.2 ppm) and C-9a (133.0 ppm) were established by long-range couplings as shown in Fig. 2. Based on the phenoxazone chromophore structure deduced by UV spectral data, the relationship between two isolated sp^2 protons 1-H (6.60 ppm) and 4-H (6.34 ppm) was assigned to para position each other as shown in Fig. 2. One of

Table 1. ¹H and ¹³C NMR assignments of 1 (500 MHz, in CDCl₃).

No.	δ_{C}	δ_{H}		
	1	105.7	6.60	(s)
	2	157.4		
	3	180.1		
	4	105.7	6.34	(s)
	4a	148.7		
	5a	142.2		
	6	116.0	7.32	(d J = 8.5 Hz)
	7	130.1	7.49	(dd J = 2.0 Hz, 8.5 Hz)
	8	139.1		
	9	127.0	7.71	(d J = 2.0 Hz)
	9a	133.0		
	10a	148.9		
	11	63.3	4.69	(s 2H)
	2-OCH ₃	56.4	3.91	(s 3H)

Fig. 2. Structure of 4-demethoxymichigazone (1) elucidated by NOE and HMBC experiments.



these isolated protons 1-H was strongly meta-coupled to a carbonyl carbon C-3 (180.1 ppm) and C-4a (148.7 ppm), and was additionally ortho-coupled to C-2 (157.4 ppm) and C-10a (148.9 ppm). Furthermore, an NOE was observed between 1-H and a methoxy proton (3.91 ppm), which was in turn long-range coupled to C-2. Thus, this methoxy residue was linked to the C-2 position. By considering the presence of the phenoxazone chromophore, C-10a was assumed to be substituted by a nitrogen atom, and the conjugated system consisting of these units was established as shown in Fig. 2. The remaining oxygen atom was linked to C-4a by elimination. Due to a low field shift of 9-H (7.71 ppm) compared with two other aromatic protons 6-H and 7-H (7.32 ppm and 7.49 ppm, respectively) on the same ring system^{9,10}, 9-H must be at a peri position to the nitrogen atom. The ¹³C chemical shifts of C-5a and C-9a also supported the connectivity between ring A and ring C as shown in Fig. 2. Even though the ¹³C chemical shifts of 2 were not fully assigned, ¹H NMR spectral data of 1 and 2 were quite resembled instead of a methoxy proton (4.05 ppm) in 1, which is replaced by a singlet aromatic proton (6.34 ppm) in 2. Thus, the structure of 1 was determined to be 4-demethoxymichigazone. Although the phenoxazone-chromophore containing compounds were reported previously and 1 was synthesized by ACHENBACH et al.⁸⁾, this paper describes the first isolation of 1 from the nature and the assignments of ¹³C chemical shifts in phenoxazone moiety.

In the evaluating system we employed^{5,11)}, **1** and **2** protected neuronal hybridoma N18-RE-105 cells from L-glutamate toxicity with ED₅₀ values 57.1 and 17.6 nm, respectively. A potent L-glutamate toxicity suppressor carquinostatin A, which was used as a positive control, showed ED₅₀ value of 12.7 nm. Since the L-glutamate toxicity in N18-RE-105 cells was reported to be caused mainly by inhibition of cystine uptake^{12,13)}, which is a precursor of the intracellular reducing substance glutathione, we examined the toxicity of a direct inhibitor of glutathione synthesis, buthionine sulfoximine (BSO), on N18-RE-105 cells. The death of N18-RE-105 cells by 500 μ M BSO was effectively prevented by **1** and **2** with

 ED_{50} values 39.8 nm and 18.5 nm, respectively. Further studies on other biological activities are now under way.

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